

REACTION OF METHANE(TRI-CHLOROACETONE) UNDER FAVORSKII CONDITIONS: A DIRECT  
ENTRY TO THE TRIASTERANE STRUCTURE

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Methane(triacetic) acid derivatives are useful starting materials for the synthesis of some unusual polycyclic cage structures.<sup>1,2</sup>

In the present communication we wish to report the reaction of methane(tri-chloroacetone)(I) -a compound easily accesible by the method we have recently described<sup>2b</sup>- under Favorskii conditions.

Dropwise addition of a solution of methane(tri-chloroacetone) in anhydrous THF to a solution of potassium ter-butoxide in ter-butyl alcohol affords, in 72% yield, a water soluble, highly crystalline, C<sub>10</sub>H<sub>10</sub>O<sub>3</sub> compound, m.p. 108°, that has been characterized by spectroscopic techniques (UV, IR, NMR and MS), as well as by mechanistic considerations and the preparation of some derivatives, as 1-hydroxymethyl-tetracyclo[3.3.1.0<sup>2,8</sup>.0<sup>4,6</sup>]nona-3,7-dione (II).

This one-step synthesis of a triasterane<sup>3</sup> derivative involves a series of intramolecular transformations (Scheme 1) that found precedents in the intermolecular base-induced reactions of mono- $\alpha$ -chloroketones:

i) proton abstraction from the more acidic CH<sub>2</sub> adjacent to both the carbonyl group and the chlorine atom<sup>4</sup>,

ii) intramolecular nucleophilic attack of the carbanion to the carbonyl group of another side chain, with simultaneous attack of the resulting alkoxide to the adjacent carbon atom and elimination of a chloride, to give an epoxide,<sup>5</sup>

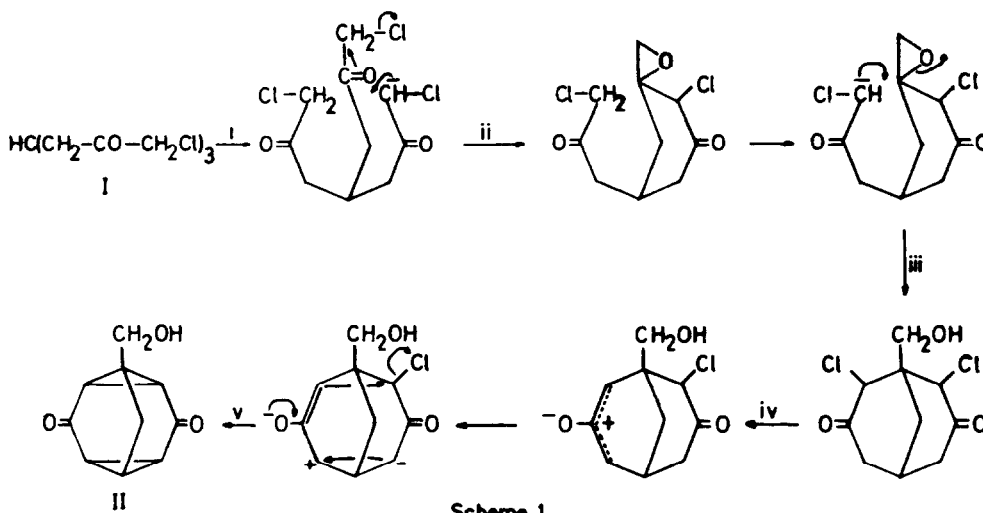
iii) intramolecular nucleophilic attack of a second carbanion to the epoxide

and formation of a bicyclic primary carbinol<sup>5</sup>,

iv) dehydrochlorination of this bicyclic carbinol to give an oxyallyl dipolar ion<sup>6</sup>, and

v) "anomalous Favorskii rearrangement" of the oxyallyl dipolar ion, induced by an intramolecular nucleophilic attack of a new carbanion, to give the triasterane II.

Confirmation of the triasterane structure for compound II could be obtained by addition of a lanthanide shift reagent, such as  $\text{Eu}(\text{fod})_3$ , to the deuteriochloroform solution, which produces a nearly first-order NMR spectrum.



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